NAME OF THE COURSE Laboratory in Chemistry Education I									
Code	PMC213		Year of st	udy	2nd year of graduate study				
Course teacher	Dr.sc. Roko Vla	adušić	Credits (ECTS)	3,0				
Associate teachers			Type of ir (number	nstruction of hours)	Р	S	V 45	Т	
Status of the course	Obligate		Percenta applicatio	ntage of 2 ation of e-learning					
COURSE DESCRIPTION									
Course objectives	The goal of the course is to prepare students for design and implementation of elementary school chemistry experiments in chemistry instruction. Special attention is paid to the development of the awareness how important role experiments do play in chemistry instruction.								
Course enrolment requirements and entry competences required for the course	Chemistry Education I obligations completed (except exam); starting competencies are related to the knowledge of chemistry and ability to work with chemicals in secure and economic way.								
Learning outcomes expected at the level of the course (4 to 10 learning outcomes)	According to the elementary school curriculum, students will be able to: - design and develop worksheets for implementation and evaluation of experimental work, - prepare and implement demonstrational and laboratory types of experiments, - create experimental situations in which pupils should make conclusions based on observations and theoretical knowledge, - perform all laboratory procedures related to experiments listed in Chemistry curriculum for elementary school, - demonstrate practical work skills and - analyse the flow and results of an experiment with focus on the cause-effect relationships								
Course content broken down in detail by weekly class schedule (syllabus)	 Substances and their properties (5 hours) Types of substances (5 hours) Air (5 hours) Vater and hydrogen (5 hours) Composition of substances (5 hours) Chemical changes (5 hours) Chemical changes (5 hours) Elements and compounds (5 hours) Metals, non-metal and salts (6 hours) Carbon and its inorganic compounds (4 hours). 								
Format of instruction	 □ lectures □ seminars and workshops ☑ exercises □ on line in entirety ☑ partial e-learning □ field work 			 ☑ independent assignments ☑ multimedia ☑ laboratory □ work with mentor □ (other) 					
Student responsibilities	To attend laboratory exercises, to design and perform experiments, to develop worksheet for experiment implementation in classroom.								
Screening student work (name the proportion of ECTS credits for each activity so that the total number of ECTS credits is equal to the ECTS	Class attendance		Research		Practical				
	Experimental work	1,5	Report		Individua laborato		0,5		
	Essay		Seminar essay		-	other)			
	Tests		Oral exam			other)			
<i>value of the course)</i> Grading and evaluating student	Written exam Preparation, im experimental ex				ents - 10				

work in class and at the final exam	experimental skills – 20 %).						
Required literature (available in the library and via other media)	Title	Number of copies in the library	Availability via other media				
	Sikirica, M. (2011). Zbirka kemijskih pokusa za osnovnu i srednju školu, Školska knjiga, Zagreb.	4					
	Chemistry textbooks applied by Ministry of science and education	5	+				
Optional literature	Sikirica, M. (2004). Metodika nastave kemije, Školska knjiga, Zagreb.						
(at the time of							
submission of study							
programme proposal)							
Quality assurance	Personal consultations, individual tasks analysis, grou	n conversatio	n institutional				
methods that	Personal consultations, individual tasks analysis, group conversation, institutional evaluation at the end of the semester.						
ensure the	evaluation at the end of the semester.						
acquisition of exit							
competences							
Other (as the							
proposer wishes to							
add)							