NAME OF THE COURSE Laboratory in Chemistry Education II									
Code	PMC214		Year of s	tudy	2nd year of gradu		luate st	uate study	
Course teacher	Dr.sc. Rok	o Vladušić	Credits (I	ECTS)	3,0				
Associate teachers				Type of instruction (number of hours)	Р	S	V 45	Т	
Status of the course	Obligate		Percenta application	ge of 5 n of e-learning					
COURSE DESCRIPTION									
Course objectives	The goal of the course is to prepare students for design and implementation of secondary schools' chemistry experiments in chemistry instruction. Special attention is paid to the development of the awareness how important role experiments do play in chemistry instruction.								
Course enrolment requirements and entry competences required for the course	Laboratory in Chemistry Education I obligations completed (except exam); starting competencies are related to the knowledge of chemistry and ability to work with chemicals in secure and economic way.								
Learning outcomes expected at the level of the course (4 to 10 learning outcomes)	According to the secondary schools curriculum, students will be able to: - design and develop worksheets for implementation and evaluation of experimental work, - prepare and implement demonstrational and laboratory types of experiments, - create experimental situations in which pupils should make conclusions based on observations and theoretical knowledge, - perform all laboratory procedures related to experiments listed in Chemistry curriculum for elementary school, - demonstrate practical work skills and - analyse the flow and results of an experiment with focus on the cause-effect relationships.								
Course content broken down in detail by weekly class schedule (syllabus)	1. Carbohydrates (5 hours) 2. Organic compounds with oxygen (5 hours) 3. Biologically important compounds (6 hours) 4. Polymers (4 hours) 5. Chemical bonding and crystals (5 hours) 6. Types of dispersion systems (5 hours) 7. Changes of energy in reaction systems (5 hours) 8. Chemical equilibrium (5 hours) 9. Electrochemistry (5 hours)								
Format of instruction	□ lectures □ seminar 図 exercise □ on line is 図 partial e □ field wos	s and workshes n entirety -learning rk	nops	 ☑ independent assignments ☑ multimedia ☑ laboratory ☐ work with mentor ☐ (other) 					
Student responsibilities	To attend laboratory exercises, to design and perform experiments, to develop worksheet for experiment implementation in classroom.)	
Screening student	Class attendance	_	Research		Practical tr	aining			
work (name the proportion of ECTS credits for each activity so that the total number of	Experimen work		Report		Individual laboratory	tasks	0,5		
	Essay		Seminar essay		Problem ar	nalysis	1		
ECTS credits is equal to the ECTS	Tests		Oral exam		(Oth				
value of the course)	Written exa		Project		(Oth	•			
Grading and evaluating student	Preparation, implementation and analysis of experiments - 100 % (or final experimental exam – 80 %; creating worksheets, structuring experiments and								

work in class and at the final exam	experimental skills – 20 %).						
Required literature (available in the library and via other media)	Title	Number of copies in the library	Availability via other media				
	Sikirica, M. (2011). Zbirka kemijskih pokusa za osnovnu i srednju školu, Školska knjiga, Zagreb.	4					
	Chemistry textbooks applied by Ministry of science and education	5					
Optional literature	Sikirica, M. (2004). Metodika nastave kemije, Školska knjiga, Zagreb.						
(at the time of							
submission of study							
programme							
proposal)							
Quality assurance	Personal consultations, individual tasks analysis, group conversation, institutional						
methods that	evaluation at the end of the semester.						
ensure the							
acquisition of exit							
competences							
Other (as the							
proposer wishes to							
add)							