

NAME OF THE COURSE		Biochemistry II				
Code	PMC225	Year of study	3rd			
Course teacher	Viljemka Bučević Popović, Assistant Professor Matilda Šprung, Assistant Professor	Credits (ECTS)	6.5			
Associate teachers	Barbara Soldo, Assistant Professor	Type of instruction (number of hours)	L	S	E	F
			30	15	45	
Status of the course	mandatory	Percentage of application of e-learning				
COURSE DESCRIPTION						
Course objectives	The objective of the course is to gain understanding of macronutrient metabolism (lipids, proteins, carbohydrates), digestion and absorption of fuel molecules, as well as their usage and storage.					
Course enrolment requirements and entry competences required for the course	The course enrolment prerequisites are General and Inorganic Chemistry and Organic Chemistry.					
Learning outcomes expected at the level of the course (4 to 10 learning outcomes)	<ol style="list-style-type: none"> 1. Demonstrate understanding of fundamental catabolic processes. Show understanding of basic anabolic processes. 2. Explain the mechanisms of regulation of metabolic processes. 3. Explain the mechanisms of storage and mobilization of fuel molecules. 4. Integrate metabolic processes at the organs level. 					
Course content broken down in detail by weekly class schedule (syllabus)	<p>Lectures:</p> <ol style="list-style-type: none"> 1. Introduction to metabolism (2 hours) 2. Glycolysis (2 hours) 3. Citric acid cycle (2 hours) 4. Respiratory chain (2 hours) 5. Oxidative phosphorylation, thermogenesis, oxidative stress (2 hours) 6. Gluconeogenesis (2 hours) 7. Pentose phosphate pathway (2 hours) 8. Glycogen metabolism and glycogen metabolism regulation (2 hours) 9. Degradation of fats and fatty acids, synthesis of ketone bodies (2 hours) 10. Synthesis of fatty acids, triacylglycerol synthesis and storage (2 hours) 11. Cholesterol (2 hours) 12. Amino acid metabolism (2 hours) 13. Hem (2 hours) 14. Nucleotide metabolism (2 hours) 15. Integration of metabolism (2 hours) <p>Seminars: Questions solving and discussion on topics that follow lectures.</p> <p>Laboratory exercises:</p> <ol style="list-style-type: none"> 1. Acid-base properties of amino acids (4 hours) 2. Time course of enzyme reaction. Enzyme kinetics (6 hours) 3. Inhibition of enzyme reaction. Activation of the enzyme reaction (6 hours) 4. Influence of temperature on enzyme activity (4 hours) 5. Protein electrophoresis (4 hours) 6. Protein separation methods. Gel-filtration (4 hours) 7. Determination of protein concentration by Bradford method (3 hours) 8. Determination of enzyme activity in natural samples: alkaline phosphatase in the sea water (6 hours) 9. Determination of bilirubin concentration (4 hours) 10. Determination of iron and iron binding capacity (4 hours) 					

Format of instruction	<input checked="" type="checkbox"/> lectures <input checked="" type="checkbox"/> seminars and workshops <input type="checkbox"/> exercises <input type="checkbox"/> <i>on line</i> in entirety <input type="checkbox"/> partial e-learning <input type="checkbox"/> field work		<input type="checkbox"/> independent assignments <input checked="" type="checkbox"/> multimedia <input checked="" type="checkbox"/> laboratory <input type="checkbox"/> work with mentor <input type="checkbox"/> (other)			
Student responsibilities						
Screening student work (<i>name the proportion of ECTS credits for each activity so that the total number of ECTS credits is equal to the ECTS value of the course</i>)	Class attendance	3	Research		Practical training	
	Experimental work		Report		(Other)	
	Essay		Seminar essay		(Other)	
	Tests	0.5	Oral exam	1.5	(Other)	
	Written exam	1.5	Project		(Other)	
Grading and evaluating student work in class and at the final exam	The written exam may be taken as two partial exams. At least 50% score is needed from each partial exam for passing grade on the written exam, followed by an oral exam.					
Required literature (available in the library and via other media)	Title			Number of copies in the library	Availability via other media	
	Jeremy M. Berg, John L. Tymoczko, Lubert Stryer, Biochemistry, 6th Ed.			8		
Optional literature (at the time of submission of study programme proposal)	Robert K. Murray, David A Bender, Kathleen M. Botham, Peter J. Kennelly, Victor W. Rodwell, P. Anthony Weil, Harper's illustrated biochemistry, 2010 Donald Voet, Judith G. Voet, Charlotte W. Pratt, Fundamentals of Biochemistry, 3rd Ed., 2005, John Wiley & Sons, Inc.					
Quality assurance methods that ensure the acquisition of exit competences	Consultations, partial examinations, student survey for subject and teacher evaluation, attendance records, quiz performance analysis, partial and final exams.					
Other (as the proposer wishes to add)						